

Palladium-Catalyzed Synthesis of New Unsaturated Exo-Enol Lactones with Potential Biological Activity.

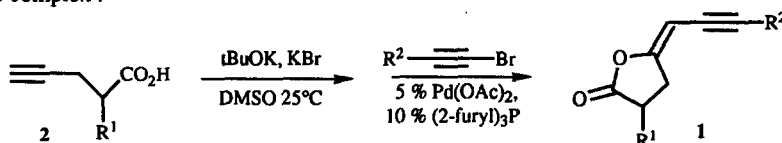
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Abstract : Various allenol lactones [5(E)-(2-allenylidene)-tetrahydro-2-furanones] have been synthesized by reacting 4-pentynoic acid and 2-alkynyl acetate in dimethyl sulfoxide in the presence of K₂CO₃, Pd(OAc)₂ as the catalyst and tri(2-furyl) phosphine as the ligand. The biological activity of these compounds as protease inhibitors has been proven.

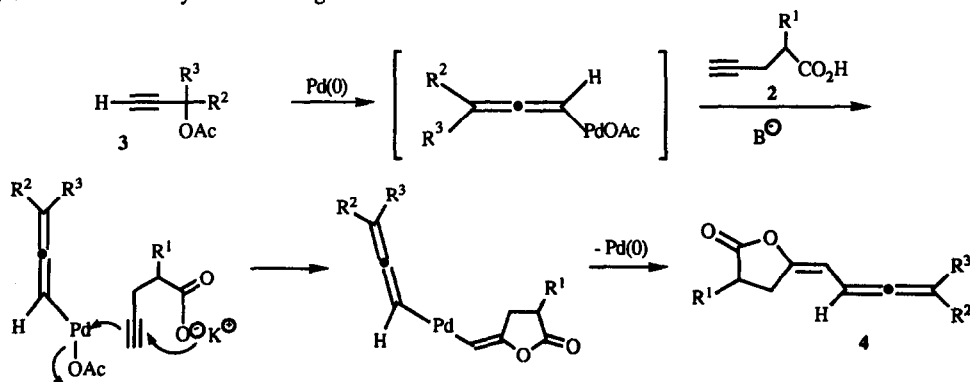
In recent years, significant attention has been focused on the synthesis of exo-enol lactones because many naturally occurring products containing this moiety possess biological activity¹.

Ynenol lactones **1** are known to display interesting biological properties mainly as suicide inhibitors of serine proteases². In an earlier publication³ we reported that these compounds are stereoselectively obtained when γ acetylenic carboxylates **2** are reacted with 1-halogeno-1 alkynes in the presence of a palladium(0) phosphine complex :



Since the reaction appeared promising for the preparation of heterocyclic compounds, we decided to examine the possibility of extending this methodology to the synthesis of allenol lactones **4** [(5E)-2 allenylidene-tetrahydro-2-furanones]. Indeed mechanistic considerations suggest that these unknown allenyl derivatives could also produce electrophilic Michael acceptors at protease active sites^{2,4}.

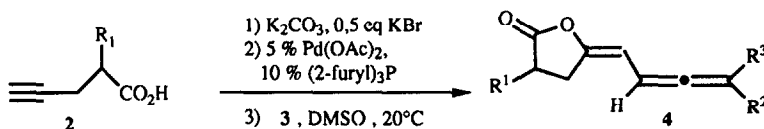
Several groups have provided many examples showing⁵ that the palladium-catalyzed reaction of propargyl derivatives offers novel synthetic methods for various allenyl compounds. Therefore a direct access to allenol lactones **4** could be the palladium catalyzed reaction of 2-alkynyl acetate **3** with pentynoic acid carboxylates **2** as shown by the following scheme :



The proposed mechanism involves an intramolecular nucleophilic attack of the carboxylate anion on the triple bond activated by a σ -allenyl palladium intermediate and subsequent reductive elimination of Pd(0) species from the resulting σ -vinyl palladium complex giving **4** and regenerating the palladium(0) complex.

The attempted reaction of 2-alkynyl acetate **3a** ($R^2=R^3=CH_3$) with the potassium carboxylate generated by the reaction of 4-pentynoic acid **2a** ($R^1=H$) with *t*BuOK under the conditions previously described for the transformation $2 \rightarrow 1^3$ was successful. We obtained the expected allenol lactone **4a** in 43% yield.

Best results were observed when K_2CO_3 was used instead of *t*BuOK to prepare the potassium carboxylate of **2a** (yield 61%). Other representative results of this reaction are shown in table 1.



Acid	Acetate	Time	Lactone	Yield % ^a
$R^1 = H$ 2a	$R^2 = R^3 = CH_3$ 3a	14 h	$R^1 = H$ $R^2 = R^3 = CH_3$	61 4a
$R^1 = H$ 2a	$R^2 = H$; $R^3 = Ph$ 3b	22 h	$R^1 = H$ $R^2 = H$; $R^3 = Ph$	50 4b
$R^1 = H$ 2a	$R^2 = R^3 = (CH_2)_5$ 3c	2 h	$R^1 = H$ $R^2 = R^3 = (CH_2)_5$	54 ^b 4c
$R^1 = CH_3$ 2b	$R^2 = R^3 = CH_3$ 3a	7 h	$R^1 = CH_3$ $R^2 = R^3 = CH_3$	62 4d

^a: the yields are based on the quantity of isolated **4** purified by flash chromatography; ^b: this reaction is carried out at 50°C.

Table 1

As noted above, allenol lactones are potential mechanism based inactivators (suicide inactivators) of serine hydrolases, since acyl transfer to the active site of the enzyme releases a dienone that can react with nucleophilic sites by the same type of Michael addition that was suggested for compounds **12**.

Our first assay using Succ-(Ala)₃-pNA as a substrate⁶ proved, as anticipated, that **4a** is an inhibitor of Human Leukocyte Elastase (E.C. 3.4.21.37) though it was less potent than compounds of type **1**. Experiments to determine the selectivity of this activity for serine proteases as well as the inhibitory capacity of diversely substituted compounds **4** are currently in progress.

References and notes

- See for example :
a. Kupchan, S.M. ; Britton, R.W. ; Ziegler, M.F. ; Gilmore, C.J. ; Restivo, R.J. and Bryan, R.F. ; *J.Amer.Chem.Soc.* **1973**, *95*, 1335. b. Niwa, M. ; Iguchi, M. and Yamamura, S. ; *Tetrahedron Lett.* **1975**, 4395. c. Kazlauskas, R. ; Murphy P.T. ; Quinn, R.J. and Wells, R.J. ; *Tetrahedron Lett.* **1977**, *37*. d. Pettus, J.A.Jr ; Wing, R.M. and Sims, J.J. ; *Tetrahedron Lett.* **1977**, 41. e. Amos, R.A. and Katzenellenbogen, J.A. ; *J.Org.Chem.* **1978**, *43*, 560. f. Yamamoto, M. ; *J.Chem.Soc. Perkin.Trans. 1.* **1982**, 582.
- Spencer, R.W. ; Tam, T.F. ; Thomas, E. ; Robinson, V.J. and Krantz, A. ; *J.Am.Chem.Soc.* **1986**, *108*, 5589.
- Bouyssi, D. ; Gore, J. and Balme, G. ; *Tetrahedron Lett.* **1992**, *33*, 2811.
- Casara, P. ; Jund, K. and Bey, P. ; *Tetrahedron Lett.* **1984**, *25*, 1891.
- a. Jeffery-Luong, T. and Linstremelle, G. ; *Tetrahedron Lett.* **1980**, *21*, 5019. b. Ruitenberg, H. ; Kleijn, H. ; Elsevier, G.J. ; Meijer, J. and Vermeer, P. ; *Tetrahedron Lett.* **1981**, *22*, 1451. c. Colas, Y. ; Cazes, B. and Gore J. ; *Tetrahedron Lett.* **1984**, *25*, 845 and *Bull.Soc. Chim.*, **1987**, 165. d. Tsuji, J. ; Watanabe, H. ; Minami, I. and Shimizu, I. ; *J.Am.Chem.Soc.* **1985**, *107*, 2196. e. Keinan, E. and Bosch, E. ; *J.Org.Chem.* **1986**, *51*, 4006. f. Mandai, T. ; Ogawa, M. ; Yamaoki, H. ; Nakata, T. ; Murayama, H. ; Kawada, M. and Tsuji, J. ; *Tetrahedron Lett.* **1991**, *32*, 3397 and references *there-in*.
- a. Bieth, J. and Wermuth, C.G. ; *Biochem.Biophys.Res.Comm.* **1973**, *53*, 383. b. Lestienne, P. and Bieth, J.C. ; *J.Biol.Chem.* **1980**, *255*, 9289.